



## Curriculum Map for Combined Science Chemistry Year 10

YEAR 10	Autumn 1	Autumn 2
<b>Topics</b>	<b>Bonding &amp; Gas Tests</b>	<b>Energy Changes</b>
<b>Substantive Knowledge – The Knowledge and Content Taught By The Teacher</b>	<ul style="list-style-type: none"><li>Basic Bonding - Ionic and Covalent</li><li>How to Test for Numerous Gases</li></ul>	<ul style="list-style-type: none"><li>Exothermic versus Endothermic</li><li>Applications of Energy Changes</li><li>Bond Energy Calculations (Higher tier only)</li></ul>
<b>Disciplinary Knowledge – The Knowledge Scientists Need So They Can Collect, Understand and Evaluate Scientific Evidence</b>	<ul style="list-style-type: none"><li>History of how the atom has changed.</li><li>History of the ideas of bonding and the theories of it.</li></ul>	<ul style="list-style-type: none"><li>History of reactions.</li></ul>
<b>Skills</b>	<ul style="list-style-type: none"><li>Using Dot-Cross Diagrams</li><li>Interpreting Compound Property Graphs</li><li>Evaluating Material Uses</li><li>Performing and Analysing a Gas Test</li></ul>	<ul style="list-style-type: none"><li>Required Practical Investigating Changes in Temperature in Reactions.</li></ul>
<b>Links To Prior Learning</b>	<ul style="list-style-type: none"><li>Atoms covered in Year 9.</li><li>Compounds introduced in Years 7 and 8 in the Materials topics.</li></ul>	<ul style="list-style-type: none"><li>Atoms covered in Year 9.</li><li>Compounds introduced in Years 7 and 8 in the Materials topics.</li><li>Bonding in Year 10 Term 1.</li></ul>
<b>Literacy/ Numeracy</b>	<ul style="list-style-type: none"><li>Interpreting compound property graphs.</li><li>Balancing equations.</li></ul>	<ul style="list-style-type: none"><li>Graphs.</li><li>Bond energy calculations.</li><li>Practical write up of investigating changes in temperature in reactions.</li></ul>
<b>Cross Curricular</b>	<ul style="list-style-type: none"><li>Technology - Use of Materials and Properties.</li><li>PSCHE - Keeping Safe with Fire and the Value of Sharing Resources.</li></ul>	<ul style="list-style-type: none"><li>Technology - Ice Packs and Heat Packs.</li><li>PSCHE - Dangers of Reactions.</li></ul>
<b>Assessment</b>	<ul style="list-style-type: none"><li>Bonding and Gas Tests Assessment</li></ul>	<ul style="list-style-type: none"><li>Energy Changes Assessment</li></ul>

<b>YEAR 10</b>	<b>Spring 1</b>	<b>Spring 2</b>
<b>Topics</b>	<b>Metals &amp; Analysis</b>	<b>Quantitative Chemistry</b>
<b>Substantive Knowledge – The Knowledge and Content Taught By The Teacher</b>	<ul style="list-style-type: none"> <li>• Metallic Bonding</li> <li>• Properties of metals</li> <li>• Analysing substances</li> <li>• Pure versus Potable Water</li> <li>• Chromatography</li> </ul>	<ul style="list-style-type: none"> <li>• Mass</li> <li>• Mole</li> <li>• Concentration</li> <li>• Relative Formula Mass</li> </ul>
<b>Disciplinary Knowledge – The Knowledge Scientists Need So They Can Collect, Understand and Evaluate Scientific Evidence</b>	<ul style="list-style-type: none"> <li>• History and ethics of clean water.</li> <li>• Properties and use of metals and how these have changed over time e.g., development of giant and nano technology.</li> </ul>	<ul style="list-style-type: none"> <li>• History of the mole.</li> </ul>
<b>Skills</b>	<ul style="list-style-type: none"> <li>• Interpreting Compound Property Graphs</li> <li>• Evaluating Material Uses</li> <li>• Treating Potable Water</li> <li>• Chromatography</li> </ul>	<ul style="list-style-type: none"> <li>• Calculations of Mass, Mole, Concentration and Relative Formula Mass</li> </ul>
<b>Links To Prior Learning</b>	<ul style="list-style-type: none"> <li>• Separating Mixtures in Years 8 and 9</li> <li>• Building on Bonding From Year 10 Term 1</li> <li>• Chromatography in Biology</li> </ul>	<ul style="list-style-type: none"> <li>• Balancing Equations in Years 8 and 9</li> <li>• Mass and Atomic Number in Years 7 and 8</li> </ul>
<b>Literacy/ Numeracy</b>	<ul style="list-style-type: none"> <li>• Doing Radiofrequency Value Calculations</li> <li>• Interpreting Compound Property Graphs</li> <li>• Development and Use of Tier 3 Key Words</li> </ul>	<ul style="list-style-type: none"> <li>• Calculations of Mass, Mole, Concentration and Relative Formula Mass</li> <li>• Conservation of Mass</li> <li>• Deciphering and Interpreting Questions and Then Doing Calculations</li> </ul>
<b>Cross Curricular</b>	<ul style="list-style-type: none"> <li>• Geography - Clean Water and Water Aid Work</li> <li>• Technology - Evaluating and Explain New Materials</li> <li>• PSCH - Clean Water for All, Health and Hygiene</li> </ul>	
<b>Assessment</b>	<ul style="list-style-type: none"> <li>• Metals and Analysis End of Topic Assessment</li> </ul>	<ul style="list-style-type: none"> <li>• Quantitative Chemistry End of Topic Assessment</li> </ul>

<b>YEAR 10</b>	<b>Summer 1</b>	<b>Summer 2</b>
<b>Topics</b>	<b>Metals &amp; Salts</b>	<b>Electrolysis &amp; Atmosphere</b>
<b>Substantive Knowledge – The Knowledge and Content Taught By The Teacher</b>	<ul style="list-style-type: none"> <li>• Reactivity Series</li> <li>• Salt - How to Make in a Variety of Ways</li> </ul>	<ul style="list-style-type: none"> <li>• Electrolysis</li> <li>• Use of Electrolysis</li> <li>• Alternative Ways to Extract Reactive Metals</li> </ul>
<b>Disciplinary Knowledge – The Knowledge Scientists Need So They Can Collect, Understand and Evaluate Scientific Evidence</b>	<ul style="list-style-type: none"> <li>• Development of the reactivity series.</li> </ul>	<ul style="list-style-type: none"> <li>• History and adaptation of technology of electrolysis to help the ever-changing world.</li> </ul>
<b>Skills</b>	<ul style="list-style-type: none"> <li>• Spotting Patterns in Chemical Reactions</li> <li>• Balancing Equations</li> <li>• Writing Correct Chemical Formulae</li> <li>• Practical - How to Make a Salt</li> </ul>	<ul style="list-style-type: none"> <li>• Electrolysis Required Practical</li> <li>• Predicting the Products of Electrolysis</li> <li>• Half Equations</li> </ul>
<b>Links To Prior Learning</b>	<ul style="list-style-type: none"> <li>• Atoms in Year 9</li> <li>• Bonding in Year 10 Term 1</li> <li>• Reactivity and Salts in Year 8</li> </ul>	<ul style="list-style-type: none"> <li>• Bonding in Year 10 Term 1</li> <li>• Electricity in Physics Year 10</li> <li>• Electricity in Physics Year 7</li> </ul>
<b>Literacy/ Numeracy</b>	<ul style="list-style-type: none"> <li>• Balancing Equations</li> <li>• Conservation of Mass</li> <li>• Extending Skills in Investigation Write-Up</li> </ul>	<ul style="list-style-type: none"> <li>• Half Equations and Balancing Them</li> <li>• Extending Skills in Investigation Write-Up</li> </ul>
<b>Cross Curricular</b>	<ul style="list-style-type: none"> <li>• PSCH E - Dangers of Fire</li> </ul>	<ul style="list-style-type: none"> <li>• PSCH E - Dangers of Electricity</li> <li>• Geography - Impacts of Human Activity on Wider Worlds</li> <li>• Physics - Electricity</li> </ul>
<b>Assessment</b>	<ul style="list-style-type: none"> <li>• Metals and Salts End of Topic Assessment</li> </ul>	<ul style="list-style-type: none"> <li>• Electrolysis and Atmosphere End of Topic Assessment</li> </ul>